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Modern Chemistry • CHAPTER 9 HOMEWORK 9-2 (pp. 280 – 282) VOCABULARY Complete each sentence. 1. ... Circle the letter of the best answer. 1. Given the balanced equation $2C_4H_{10} + 13O_2 \rightarrow 8CO_2 + 10H_2O$, how many moles of CO_2 are produced when 14.9 g of O_2 are used? a. 0.29 mol b. 3.49 mol

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Section 2 Answers Modern Chemistry Chapter 9 Stoichiometry Chapter 9 –
Stoichiometry Review #1 – #18, #31, & #38 Answers . 38. To ensure that all
magnesium is converted to MgO, I would use pure oxygen, not air, to carry out the
reaction, because Mg could react with N₂ in air to form Mg₃N₂. The pure oxygen

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should be in excess. 5. a. 50 mol HI Page 6/24

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Answer Key Chapter 9 - Chemistry 2e | OpenStax 1. The cutting edge of a knife that has been sharpened has a smaller surface area than a dull knife. Since pressure is force per unit area, a sharp knife will exert a higher pressure with the same amount of force and cut through material more effectively.

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