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which kinetic energy may be transferred
between.... a measure of the average
kinetic energy of the particles in a....

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13. Pressure is defined as force per unit a.

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13.78 kPa 6. Find the partial pressure of
carbon dioxide in a gas mixture with a
total pressure of 30.4 kPa if the partial

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pressures of the other two gases in the mixture are 16.5 kPa and 3.7 kPa. 30.4 kPa 16.5 kPa 3.7 kPa 10.2 kPa Solutions Manual Chemistry: Matter and Change • Chapter 12 237

States of Matter

Notice in particular the curves for NH_4NO_3 and CaCl_2 . The dissolution of

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ammonium nitrate in water is endothermic ($\Delta H_{\text{soln}} = +25.7 \text{ kJ/mol}$), whereas the dissolution of calcium chloride is exothermic ($\Delta H_{\text{soln}} = -68.2 \text{ kJ/mol}$), yet Figure 13.4.1 shows that the solubility of both compounds increases sharply with increasing temperature. In fact, the magnitudes of the changes in both enthalpy ...

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is constant in the following problems 11
A sample of air in a syringe exerts a
pressure of 102 atm at 220°C The

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